

RECEIVED

OCT 10 1992



Europäisches Patentamt
European Patent Office
Office européen des brevets



(11) Publication number: **0 524 731 A1**

(12)

EUROPEAN PATENT APPLICATION

(21) Application number: **92305885.3**

(51) Int. Cl.⁵: **C08L 69/00, C08L 83/10,
// (C08L69/00, 83:10)**

(22) Date of filing: **25.06.92**

(30) Priority: **01.07.91 US 724023**

(43) Date of publication of application:
27.01.93 Bulletin 93/04

(84) Designated Contracting States:
DE ES FR GB IT NL

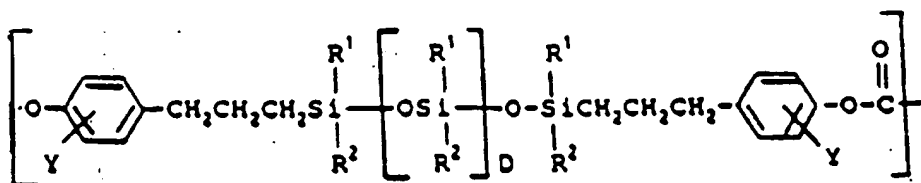
(71) Applicant: **GENERAL ELECTRIC COMPANY**
1 River Road
Schenectady, NY 12345 (US)

(72) Inventor: **Hoover, James Franklin**
R.R.13, Box 150B, Copperline Road
Evansville, Indiana 47712 (US)

(74) Representative: **Pratt, Richard Wilson et al**
London Patent Operation G.E. Technical
Services Co. Inc. Essex House 12/13 Essex
Street
London WC2R 3AA (GB)

(54) Polymer blends of polycarbonate-polysiloxane block copolymers with polycarbonates and polyester carbonate copolymers.

(57) New thermoplastic blends are provided, comprising (A) a block copolymer having a polysiloxane block of the structure:



where R¹, R², Y and D are as herein defined, and a polycarbonate block, and (B), an aromatic carbonate polymer. These blends have processing, low temperature impact strength and flammability advantages.

EP 0 524 731 A1

Jouve, 18, rue Saint-Denis, 75001 PARIS

BACKGROUND OF THE INVENTION

Field of the Invention

5 This invention relates to thermoplastic blends of polysiloxane-polycarbonate block copolymers with carbonate polymers, useful as engineering thermoplastics with improved impact strength and processability.

Brief Description of Related Art

10 Condensation copolymers having polysiloxane and polycarbonate blocks are known. Representative of such polymers are those disclosed by Schmidt et al., U. S. Pat. 4,681,922 (July 1987), Vaughn, U. S. Pat. 3,189,662 (June 1965) and Vaughn, U. S. Pat. 3,419,634 (Dec. 1968).

The above-described copolymers, while useful, require that a number of different grades, differing in siloxane level, must be manufactured to meet the various market requirements for differing balances of low and high temperature properties. It would be desirable to have means for meeting these diverse requirements by manufacturing only a single grade of siloxane-modified polycarbonate copolymer.

A highly advantageous feature of the invention is the finding that certain siloxane-carbonate block copolymers can be blended with various proportions of aromatic carbonate polymers to prepare thermoplastic molding compositions capable of meeting the wide range of requirements for high and low temperature properties. These blends also exhibit other useful advantages such as an ease of achieving flame retardancy. By varying the selection of the carbonate polymer and the proportions of blend ingredients, a wide range of properties may be obtained, starting with a single siloxane-carbonate block copolymer.

SUMMARY OF THE INVENTION

25

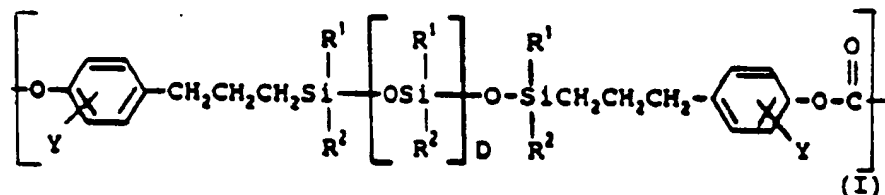
The invention comprises a thermoplastic composition which comprises a blend of:

(A) about 5 to 50 weight percent of a thermoplastic block copolymer comprising:

(1.) about 1 to about 50 % by weight of a siloxane from recurring polysiloxane blocks of the formula:

30

35



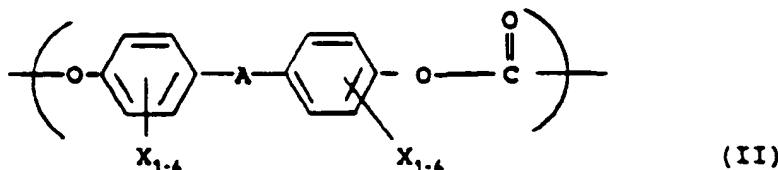
40

where R¹ and R² are each independently selected from hydrogen, hydrocarbyl and halogen-substituted hydrocarbyl, (preferably R¹ is methyl and R² is methyl or phenyl); D is an integer of from about 10 to about 120, preferably about 40-60, Y is hydrogen or alkoxy (and where alkoxy, preferably methoxy); and

(2.) about 50 to about 99 % by weight of the block copolymer of a polycarbonate-block comprising units of the formula:

45

50



55

wherein A is a divalent hydrocarbon radical containing from 1 to about 15 carbon atoms: a substituted divalent hydrocarbon radical containing from 1 to about 15 carbon atoms and substituent groups such as halogen: -S- : -SS- : -S(O)- : -S(O)₂- : -O- : or -C- : wherein each X is independently selected from the group consisting of hydrogen, hydrocarbyl such as an alkyl group of from 1 to about 8 carbon atoms,

an aryl group of from 6-18 carbon atoms, an aralkyl group of from 7 to about 14 carbon atoms, an alkaryl group of from 7 to about 14 carbon atoms, an alkoxy group of from 1 to about 8 carbon atoms, or an aryloxy group of from 6 to 18 carbon atoms and halogen; and

B. From 50 to 95 weight percent of an aromatic carbonate polymer.

It is preferred that the proportions of (A) and (B) are such that the siloxane moieties comprise about 4 to 8 weight percent of the total blended composition.

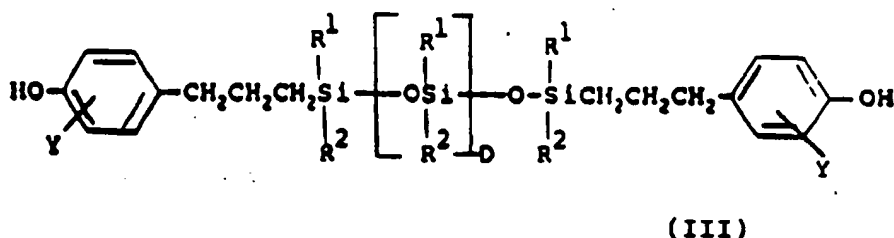
The term "hydrocarbyl" as used herein means the monovalent moiety obtained upon removal of a hydrogen atom from a parent hydrocarbon. Representative of hydrocarbyl are alkyl of 1 to 25 carbon atoms, inclusive such as methyl, ethyl, propyl, butyl, pentyl, hexyl, heptyl, octyl, nonyl, undecyl, decyl, dodecyl, octadecyl, nonadecyl, eicosyl, heneicosyl, docosyl, tricosyl, tetracosyl, pentacosyl and the isomeric forms thereof; aryl of 6 to 25 carbon atoms, inclusive, such as phenyl, tolyl, xylol, naphthyl, biphenyl, tetraphenyl and the like; aralkyl of 7 to 25 carbon atoms, inclusive, such as benzyl, phenethyl, phenpropyl, phenbutyl, phenhexyl, naphthoctyl and the like; cycloalkyl of 3 to 8 carbon atoms, inclusive, such as cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cycloheptyl, cyclooctyl and the like.

DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENTS OF THE INVENTION

The thermoplastic block copolymer (A) may be formed by the interfacial polymerization of a carbonate precursor such as phosgene with

(1) a dihydric phenol of the formula IV shown below:

(2) a siloxane of the formula:

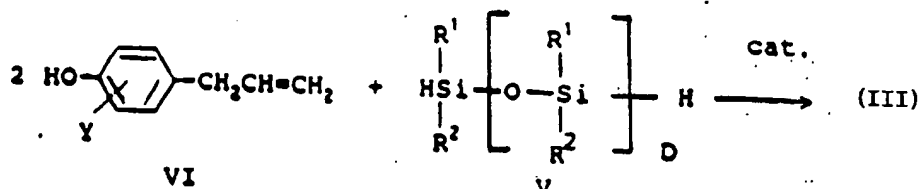


where R^1 and R^2 , D and Y are as defined above:

(3) a chain length limiting amount of a monohydric phenol, preferably phenol itself.

The procedure is the conventional one used to prepare polycarbonates and polyester-carbonates.

The siloxanes (III) may be characterized as bisphenolsiloxanes. The preparation of these bisphenolsiloxanes is accomplished by the addition of a polydiorganosiloxane (V) to a phenol (VI) containing an alkenyl substituent, according to the schematic formula:



wherein R¹, R², Y and D are as defined above.

The essential features of the process are described by Vaughn, U. S. Pat. 3,419,635 (Dec. 1968), which is incorporated by reference. For instance, the process is exemplified in example 8 of this Vaughn patent which describes the addition of a hydrogen-terminated polydimethylsiloxane to an allylphenol in the presence of a catalytic amount of chloroplatinic acid-alcohol complex at 90-115°C.

The aromatic carbonate polymers B, employed in the blends of the invention encompass polycarbonates as well as polyester-carbonates. The method of preparation of polycarbonates and polyester carbonates by interfacial polymerization are well known; see for example the details provided in the U.S. Patents 3,028,365; 3,334,154; 3,275,601; 3,915,926; 3,030,331; 3,169,121; 3,027,814; and 4,188,314, all of which are incorpo-

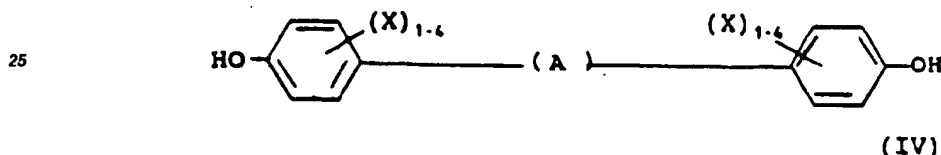
rated herein by reference thereto.

In general, the method of interfacial polymerization comprises the reaction of a dihydric phenol with a carbonyl halide (the carbonate precursor).

Although the reaction conditions of the preparative processes may vary, several of the preferred processes typically involve dissolving or dispersing the diphenol reactants in aqueous caustic, adding the resulting mixture to a suitable water immiscible solvent medium and contacting the reactants with the carbonate precursor, such as phosgene, in the presence of a suitable catalyst and under controlled pH conditions. The most commonly used water immiscible solvents include methylene chloride, 1,2-dichloroethane, chlorobenzene, toluene, and the like.

The catalyst employed accelerates the rate of polymerization of the dihydric phenol reactant with the carbonate precursor. Representative catalysts include but are not limited to tertiary amines such as triethylamine, quaternary phosphonium compounds, quaternary ammonium compounds, and the like. The preferred process for preparing polycarbonate resins of the invention comprises a phosgenation reaction. The temperature at which the phosgenation reaction proceeds may vary from below 0°C, the above 100°C. The phosgenation reaction preferably proceeds at temperatures of from room temperatures (25°C) to 50°C. Since the reaction is exothermic, the rate of phosgene addition may be used to control the reaction temperature. The amount of phosgene required will generally depend upon the amount of the dihydric phenol and the amount of any dicarboxylic acid also present.

The dihydric phenols employed are known, and the reactive groups are the two phenolic hydroxyl groups. Some of the dihydric phenols are represented by the general formula:



wherein A and X are as defined above.

Typical of some of the dihydric phenols that can be employed in the practice of the present invention are bis-phenols such as (4-hydroxyphenyl)methane, 2,2-bis(4-hydroxyphenyl)propane (also known as bisphenol-A), 2,2-bis(4-hydroxy-3,5-dibromophenyl)propane; dihydric phenol ethers such as bis(4-hydroxyphenyl) ether, bis(3,5-dichloro-4-hydroxyphenyl) ether; dihydroxydiphenyls such as p,p'-dihydroxydiphenyl, 3,3'-dichloro-4,4'-dihydroxydiphenyl; dihydroxyaryl sulfones such as bis(4-hydroxyphenyl) sulfone, bis(3,5-dimethyl-4-hydroxyphenyl) sulfone, dihydroxybenzenes such as resorcinol, hydroquinone, halo- and alkyl-substituted dihydroxybenzenes such as 1,4-dihydroxy-2,5-dichlorobenzene, 1,4-dihydroxy-3-methylbenzene; and dihydroxydiphenyl sulfides and sulfoxides such as bis(4-hydroxyphenyl) sulfide, bis(4-hydroxyphenyl) sulfoxide and bis(3,5-dibromo-4-hydroxyphenyl) sulfoxide. A variety of additional dihydric phenols are available and are disclosed in U.S. Pat. Nos. 2,999,835; 3,028,365 and 3,153,008; all of which are incorporated herein by reference. It is, of course, possible to employ two or more different dihydric phenols or a combination of a dihydric phenol with glycol.

The carbonate precursor can be either a carbonyl halide, a diarylcarbonate or a bishaloformate. The carbonyl halides include carbonyl bromide, carbonyl chloride, and mixtures thereof. The bishaloformates include the bishaloformates of dihydric phenols such as bischloroformates of 2,2-bis(4-hydroxyphenyl)propane, 2,2-bis(4-hydroxy-3,5-dichlorophenyl)propane, hydroquinone, and the like, or bishaloformates of glycols such as bishaloformates of ethylene glycol, and the like. While all of the above carbonate precursors are useful, carbonyl chloride, also known as phosgene, is preferred.

The composition blends of the invention may be modified by the addition of additives known to the art of plastics compounding. Such additives can include for example fillers (such as clay or talc), reinforcing agents (such as glass fibers), impact modifiers, other resins, antistats, plasticizers, flow promoters and other processing aids, stabilizers, colorants, mold release agents, other flame retardants, ultraviolet screening agents, and the like.

In addition to the advantages of improved processing and improved low temperature properties, another important feature of the resin blend compositions of the invention is their ability to produce articles which display good flame ignition resistance and flame retardancy, while at the same time having good impact strength over a wide temperature range. It has hitherto been difficult to achieve both desirable properties at the same time

without a degree of compromise. For the purpose of achieving an enhanced degree of flame retardancy, the blends of the invention may be further blended with alkali metal or alkaline earth metal salts of organic sulfonates which may be halogenated, with polycarbonates made with halogenated bisphenols, with halogenated organic molecules, organophosphorus compounds, halogen-phosphorus compounds, polytetrafluoroethylene, fumed silica, and other materials which provide flame ignition resistance (or flame retardancy) to the resin composition and to molded articles made therefrom. Specially preferred compositions of the invention are those that contain an effective flame retardant amount, typically from 0.3 to 0.6 % sulfonate or from 0.5 to 2.0% of halogenated compounds.

The invention will be better understood with reference to the following examples, which are presented for purposes of illustration rather than for limitation, and are set forth the best mode contemplated for carrying out the invention.

Preparation 1

Representative preparation of eugenol capped polydimethylsiloxane fluid:

Octamethylcyclotetrasiloxane (8.3 kg, 28.0 moles), tetramethyldisiloxane (330 g, 2.46 moles) and Filtrol 20 (86 g, 1% by weight, Harshaw/Filtrol Clay Products) were combined in a 12 L flask and heated to 45 C for 2 hours. The temperature was raised to 100 C and the mixture was rapidly agitated for 5 hours. The mixture was allowed to cool then filtered through a plug of Celite filtering aid. To the crude product was added a mixture of eugenol (774 g, 4.72 moles) and Karstedt's platinum catalyst (1.57 g, 10 ppm Pt) at a rate of 40 g/minute. Reaction completion was monitored by the disappearance of the siloxane hydrogen in the FTIR spectrum. The reaction product was stripped of volatiles using a falling thin film evaporator operating at 200 C and 1.5 torr. The isolated material was a light brown oil with a viscosity of 100 cstokes at 25 C and a degree of polymerization of 49 siloxane units. The material was used without further purification.

Preparation 2

Representative preparation of the copolymer bisphenol-A (90.7 kg, 398 moles), the D49 eugenol capped polydimethylsiloxane fluid (27.1 kg, 1.5 moles, 20 percent by weight siloxane), triethylamine (1.1 L), phenol (1.4 kg, 14.9 mole) and sodium gluconate (150 g) were combined with water (200 L) and dichloromethane (250 L) in the 1000 L reactor. The biphasic mixture was stirred vigorously and brought to pH 10 by the addition of 50% aqueous sodium hydroxide solution. Phosgene (45.3 kg, 458 moles) was added over the course of 30 minutes with the reaction pH maintained in the range of 10.0 to 10.5. After purging the reaction mixture with nitrogen, additional dichloromethane (200 L) was added and the organic phase was separated from the brine layer by liquid/liquid centrifugation. The organic phase was worked up by washing with 1% aqueous hydrochloric acid solution and water by means of phase to phase contact followed by liquid/liquid centrifuge separation. The resin was isolated as white granules (95 kg) by steam precipitation from the dichloromethane solution. The material was characterized by standard analytical methods. IV 0.47; Gel Permeation Chromatography (polystyrene standard using a K factor to polycarbonate molecular weights) Mw 24023, Mn 10529, Disp 2.28; Melt flow 13 g/10 min. Siloxane content by ¹H NMR was 19 wt%.

By similar methods, various other resin compositions are prepared and evaluated with different siloxane degree of polymerization and overall loading of siloxane. Several representative examples of blends of the polysiloxane copolycarbonate are given below.

Materials were compounded with a phosphite stabilizer (0.05 wt.%) and tumbled in a stainless steel can prior to extrusion on a Werner and Pfleiderer 28 or 30 mm co-rotating twin screw extruder operating at 550 to 570 F barrel temperatures. Test specimens were prepared by injection molding at 570 F melt temperature and 180 F mold temperature.

Notched IZOD impact of single gated bars and unnotched IZOD impact of double gated bars were determined according to ASTM D-256. Flexural yield and flexural modulus were determined according to ASTM D-790. Tensile yield, tensile break and tensile elongation were determined according to ASTM D-638. Melt flow rating was determined by modified ASTM D-1238. Ignition resistance was determined by UL94 test.

Example 1. The following compositions were prepared as a means of demonstrating the invention.

A composition of the present invention, 1, was prepared: 25 parts copolycarbonate of bisphenol-A and eugenol capped polydimethylsiloxane with a siloxane degree of polymerization equal to 64 units having 20 weight percent siloxane, 75 parts LEXAN 140 grade polycarbonate resin and appropriate stabilizers.

A composition of the present invention, 2, was prepared: 25 parts copolycarbonate of bisphenol-A and eugenol capped polydimethylsiloxane with a siloxane degree of polymerization equal to 64 units having 20 weight percent siloxane, 75 parts LEXAN-SP 100 grade copolyester carbonate resin (prepared in accordance with For-

tana, et. al. in U. S. Pat. 4,983,706 using 10 weight percent dodecanedioic acid) and appropriate stabilizers.

A composition of the present invention, 3, was prepared: 25 parts copolycarbonate of bisphenol-A and eugenol capped polydimethylsiloxane with a siloxane degree of polymerization equal to 64 units having 20 weight percent siloxane, 75 parts LEXAN-PPC 345 grade aromatic copolyester carbonate resin (prepared in accordance with Quinn in U. S. Pat. 4,238,596 using 80 weight percent isophthaloyl chloride/terephthaloyl chloride in a 93 to 7 ratio) and appropriate stabilizers.

The materials were compounded, extruded and molded as described above:

Property	Invention 1	Invention 2	Invention 3
Tg (°C)	145	130	185
Melt flow rating (g/10 min)	5	7	< 2
Viscosity @ 500 sec ⁻¹ (Pa-sec)	1042	736	2110
Viscosity @ 1000 sec ⁻¹ (Pa-sec)	676	495	1189
Viscosity @ 5000 sec ⁻¹ (Pa-sec)	191	154	240
125 mil double gated IZOD (ft-lbs/in)	26*	32*	11†
125 mil notched IZOD @ -10°C	-	-	8*
125 mil notched IZOD @ -50°C	11*	13*	6†
250 mil notched IZOD @ -40°C	8*	5†	6†
craze resistance to diacetone alcohol (sec)	65	15	> 1000

* indicates ductile failure;

† indicates brittle failure.

1. Time to initial craze formation after exposing a 125 mil thickness bar in a 3400 psi constant radius strain jig to solvent.

These data indicate the range of properties available from the blend compositions of the invention.

Example 2. Control composition, 4, was prepared: 91.4 parts LEXAN 140 grade polycarbonate resin (GE Plastics), 4.3 parts B56 impact modifier (a methyl methacrylate butadiene styrene core shell copolymer from Kameka Texas Corporation), 4.3 parts of a copolycarbonate of bisphenol-A and 43 weight percent bisphenol-A capped polydimethylsiloxane with a degree of polymerization of 10 (made by GE in accordance with Merritt, U. S. Pat. 3,832,419), 0.45 parts potassium diphenylsulfone-3-sulfonate (flame retardant), 2.0 parts TIONA RCL-69 titanium dioxide (made by SCM) and appropriate stabilizers.

A composition of the invention, 5, was prepared: 25 parts copolycarbonate of bisphenol-A and eugenol capped polydimethylsiloxane with a siloxane degree of polymerization equal to 100 units having 20 weight percent siloxane, 75 parts LEXAN 120 grade polycarbonate resin, 0.45 parts potassium diphenylsulfone-3-sulfonate (flame retardant), 2.0 parts TIONA RCL-69 titanium dioxide (made by SCM) and appropriate stabilizers.

The materials were compounded, extruded and molded as described above:

	Control 4	Invention 5
Property		
Melt flow rating (g/10 min)	11	8.5
Part melt flow rating (g/10 min)	11	8.5
Tensile at break (psi)	9400	9200
Tensile at yield (psi)	8200	8100
Tensile elongation (%)	104	96
Flexural modulus (psi)	302300	311600
Flexural yield (psi)	11900	12100
125 mil notched IZOD (ft-lbs/in)	13*	16*
125 mil notched IZOD @ -50 C	11*	12*
250 mil notched IZOD (ft-lbs/in)	12*	13*
UL94 at 125 mil (rating)	B	V0
* indicates ductile failure		

These data show that at approximately similar impact performance, the composition of the invention has superior ignition resistance as measured by UL94.

Example 3. A control composition, 6, was prepared: 100 parts LEXAN 140 grade polycarbonate resin, 0.45 parts potassium diphenylsulfone-3-sulfonate, 1.0 part copolycarbonate of bisphenol-A and tetrabromobisphenol-A, 0.60 parts blend of polytetrafluoroethylene (20%) with bisphenol-A polycarbonate (80%), 2.0 parts TIONA RCL-69 titanium dioxide (made by SCM) and appropriate stabilizers.

A composition of the invention, 7, was prepared: 25 parts copolycarbonate of bisphenol-A and eugenol capped polydimethylsiloxane with a siloxane degree of polymerization equal to 100 units having 20 weight percent siloxane, 75 parts LEXAN 120 grade polycarbonate resin, 0.45 parts potassium diphenylsulfone-3-sulfonate, 1.0 part copolycarbonate of bisphenol-A and tetrabromobisphenol-A, 0.60 parts blend of polytetrafluoroethylene (20%) with bisphenol-A polycarbonate (80%), 2.0 parts TIONA RCL-69 titanium dioxide (made by SCM) and appropriate stabilizers.

A second composition of the invention, 8, was prepared: 40 parts copolycarbonate of bisphenol-A and eugenol capped polydimethylsiloxane with a siloxane degree of polymerization equal to 50 units having 20 weight percent siloxane, 60 parts LEXAN 120 grade polycarbonate resin, 0.45 parts potassium diphenylsulfone-3-sulfonate, 1.0 part copolycarbonate of bisphenol-A and tetrabromobisphenol-A, 1.0 parts blend of polytetrafluoroethylene (20%) with bisphenol-A polycarbonate (80%), 2.0 parts TIONA RCL-69 titanium dioxide (made by SCM) and appropriate stabilizers.

A third composition of the invention, 9, was prepared: 12 parts copolycarbonate of bisphenol-A and eugenol capped polydimethylsiloxane with a siloxane degree of polymerization equal to 10 units having 43 weight percent siloxane, 88 parts LEXAN 120 grade polycarbonate resin, 0.45 parts potassium diphenylsulfone-3-sul-

fonate, 1.0 part copolycarbonate of bisphenol-A and tetrabromobisphenol-A, 1.2 parts blend of polytetrafluoroethylene (20%) with bisphenol-A polycarbonate (80%), 2.0 parts TIONA RCL-69 titanium dioxide (made by SCM) and appropriate stabilizers.

The materials were compounded, extruded and molded as described above:

5

10

15

20

25

30

35

40

45

50

55

Property	Control 6	Invention 7	Invention 8	Invention 9
Overall wt% siloxane	0	5	8	5
Siloxane degree of polymerization	-	100	50	10
Melt flow rate (g/10 min)	8.5	9.5	13	13
Part melt flow rate (g/10 min)	7	10	13.5	13
Tensile at break (psi)	10300	8400	7500	8200
Tensile at yield (psi)	9400	8100	7900	8000
Tensile elongation (%)	98	79	65	74
Flexural modulus (psi)	328900	296100	292800	287900
Flexural yield	14000	11700	11500	11300
125 mil notched IZOD (ft-lbs/in)	15*	12*	13*	13*
125 mil notched IZOD @ -30°C	2†	11*	10*	9*
250 mil notched IZOD (ft-lbs/in)	2†	8*	10*	11*
at 60 mil (rating) (UL94)	V0	V0	V0	V0
flame out time (sec) (Average)	2	2	4	1
125 mil (rating)(5V)	Failed	Passed	Passed	Passed

* indicates ductile failure; † indicates brittle failure

These data show that at comparable ignition resistance performance, the compositions of the invention are superior in impact performance throughout a range of compositions of siloxane degree of polymerization and loading.

Example 4. Control composition, 10, was prepared: 90 parts LEXAN 140 grade polycarbonate resin (GE Plastics), 10 parts chopped glass fibers (Owens Corning Fiberglass 415), 0.45 parts potassium diphenylsulfone-3-sulfonate (flame retardant), 2.0 parts TIONA RCL-69 titanium dioxide (made by SCM) and appropriate stabilizers.

A composition of the invention, 11, was prepared: 25 parts copolycarbonate of bisphenol-A and eugenol capped polydimethylsiloxane with a siloxane degree of polymerization equal to 50 units having 20 weight percent siloxane, 65 parts LEXAN 140 grade polycarbonate resin, 10 parts chopped glass fibers (Owens Corning Fiberglass 415), 0.45 parts potassium diphenylsulfone-3-sulfonate (flame retardant), 2.0 parts TIONA RCL-69 titanium dioxide (made by SCM) and appropriate stabilizers.

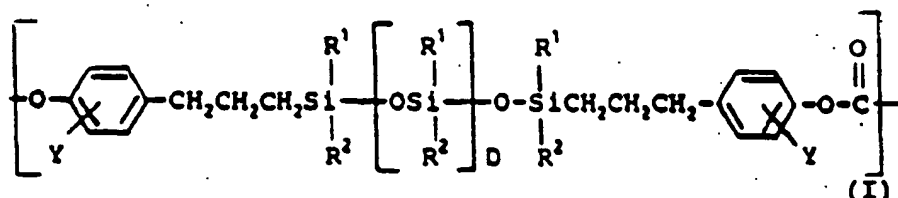
The materials were compounded, extruded and molded as described above:

	Control 10	Invention 11
Property		
Melt flow rating (g/10 min)	8	7
Part melt flow rating (g/10 min)	8	7
Tensile @ break (psi)	7100	6300
Tensile @ yield (psi)	8800	7200
Tensile elongation (%)	15	15
Flex modulus (psi)	397100	326800
Flex yield	14100	11600
125 mil notched IZOD (ft-lbs/in)	2+	5*
250 mil notched IZOD (ft-lbs/in)	1+	4*
UL94 @ 90 mil (rating)	V2	V0
(* indicates ductile failure; + indicates brittle failure)		

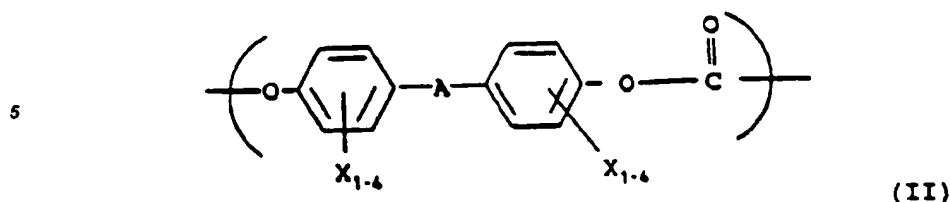
These data indicate that at equal glass fiber loading for dimensional reinforcement the compositions of the invention show superior ignition resistance and impact performance.

Claims

1. A thermoplastic composition which comprises a blend of :
 - (A) about 5 to 50 weight percent of a thermoplastic block copolymer comprising :
 - (1.) about 1 to about 50% by weight of siloxane comprising polysiloxane blocks of the formula :



where R¹ and R² are each independently selected from hydrogen, hydrocarbyl and halogen-substituted hydrocarbyl, D is an integer of from about 10 to about 120; Y is hydrogen or alkoxy; and
(2.) about 50 to about 99% by weight of the block copolymer of a polycarbonate-block comprising recurring units of the formula :



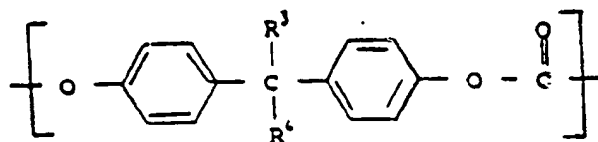
10

wherein A is a divalent hydrocarbon radical containing from 1 to about 15 carbon atoms; a substituted divalent hydrocarbon radical containing from 1 to about 15 carbon atoms and substituent groups such as halogen; -S-; -SS-; -S(O)-; -S(O)₂-; -O-; or -C-; each X is independently selected from the group consisting of hydrogen, halogen, and a monovalent hydrocarbon radical, and

15 (B) about 50 to 95 weight percent of an aromatic carbonate polymer.

2. The blend of Claim 1 wherein said aromatic carbonate polymer B. is a polycarbonate comprising repeating units of the formula :

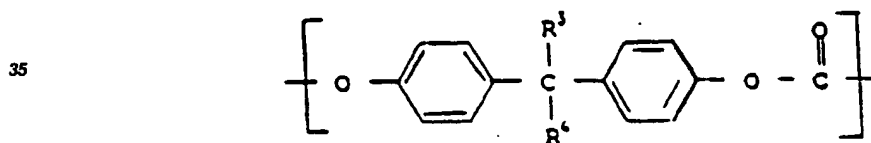
20



25

where R³ and R⁴ are independently selected from hydrogen, hydrocarbyl and halogen substituted hydrocarbyl.

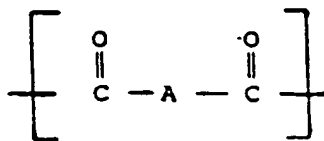
3. The blend of Claim 1 wherein said aromatic carbonate polymer is a polyester-carbonate comprising recurring carbonate units of the formula :



40

where R³ and R⁴ are selected from the group consisting of hydrogen, hydrocarbyl and halogen-substituted hydrocarbyl, copolycondensed with 1 to 20 mole %, relative to total carbonate and ester units of the formula :

45



50

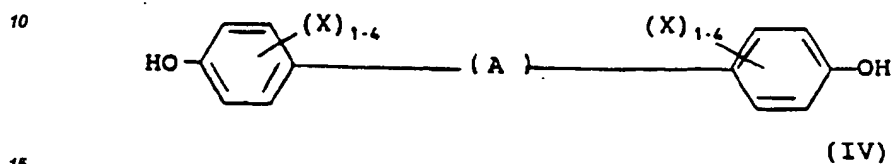
where A is alkylene of about 6 to about 18 carbon atoms or phenylene.

4. The blend of Claim 3 wherein said alkylene is straight chain alpha-omega alkylene.
- 55 5. The blend of Claim 4 wherein said alkylene is hexamethylene.
6. The blend of any preceding claim wherein R¹, R², R³ and R⁴ are methyl and Y is methoxy.

7. The blends of any preceding claim wherein D is about 40-60.
8. A process for the preparation of a thermoplastic composition of any preceding claim comprising the steps of

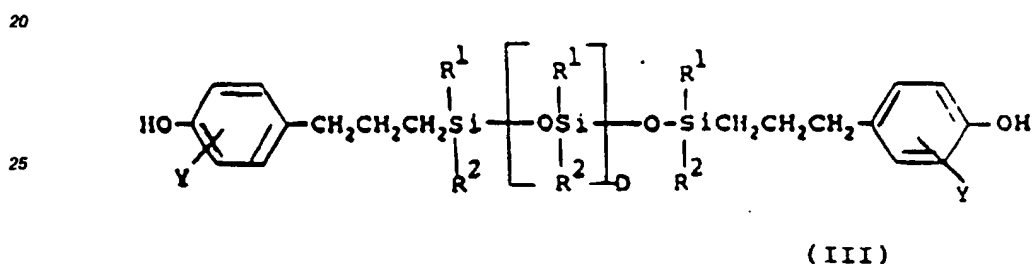
5 (1) forming a block copolymer by introducing a carbonate precursor into a stirred two phase mixture comprising

(a) a dihydric phenol of the formula



wherein A and X are as defined in Claim 1

(b) a bisphenol siloxane of the formula



30 wherein R¹, R², D and Y are as defined in Claim 1, in the presence of an effective molecular weight modifying amount of a monohydric phenol and sufficient aqueous alkali to maintain an alkaline pH and a substantially water-immiscible solvent and

(2) blending said block copolymer with an aromatic polycarbonate or an aromatic copolymer carbonate



European Patent
Office

EUROPEAN SEARCH REPORT

Application Number

EP 92 30 5885

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. CL.5)
P, X	CHEMICAL ABSTRACTS, vol. 116, no. 12, 23 March 1992, Columbus, Ohio, US; abstract no. 107621U, UNEDA ET AL.: 'Glass fiber - reinforced polycarbonate resin compositions.' page 58 ; * abstract * & JP-A-3 243 656 (IDEMITSU PETROCHEMICAL CO., LTD.) 30 October 1991	1, 2, 7, 8	C08L69/00 C08L83/10 //(C08L69/00, 83: 10)
P, Y	EP-A-0 434 848 (IDEMITSU PETROCHEMICAL CO. LTD.) * claims 1-3 * * page 4, line 21 - line 54 * * page 5, line 27 - line 50 * * page 6, line 29 - line 30 * & WO-A-9 100 885	1, 2, 8	
Y		1, 2, 8	
X	EP-A-0 376 052 (IDEMITSU PETROCHEMICAL CO. LTD.)	1, 2, 8	
Y	* claims 1-3, 7, 8 * * page 3, line 25 - line 40 * * page 5, line 35 - line 48 * * page 7, lines 19 - 24, 44 - 52 *	1, 2, 8	TECHNICAL FIELDS SEARCHED (Int. CL.5)
A	FR-A-1 507 429 (GENERAL ELECTRIC COMPANY) * page 2, left column, paragraph 4; example 7*	1, 2, 6-8	C08L
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 11 SEPTEMBER 1992	Searcher KANETAKIS I.
<p>CATEGORY OF CITED DOCUMENTS</p> <p>X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document</p> <p>T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application I : document cited for other reasons * : member of the same patent family, corresponding document</p>			